

Group 16

Design and Development of Optimized Flow Channels for an Alkaline Membrane Fuel Cell Educational Kit

Advisor/Sponsor:
Dr. Juan Ordonez

Oluwafemi Ojo

Tristan Walter

Terry Grandchamps

Trevor Gwisz



Presentation Overview

- ▶ Project Goal Statement
- ▶ Brief Re-Introduction
- ▶ Current Design
- ▶ Functional Analysis
- ▶ Failure Analysis
- ▶ Safety Analysis
- ▶ Commercialization
- ▶ Plans for the Future

Goal Statement

- ▶ “Deliver a functioning educational alkaline membrane fuel cell kit that demonstrates the effects of flow configurations on the fuel cell’s performance by the end of spring 2017 semester”

Re-Introduction to AMFC Operation

- ▶ Converts chemical energy into electric potential energy
- ▶ Requires an electrolyte solution, hydrogen gas, and oxygen gas for operation
- ▶ Generates pure water and electricity
- ▶ No greenhouse gas emitted

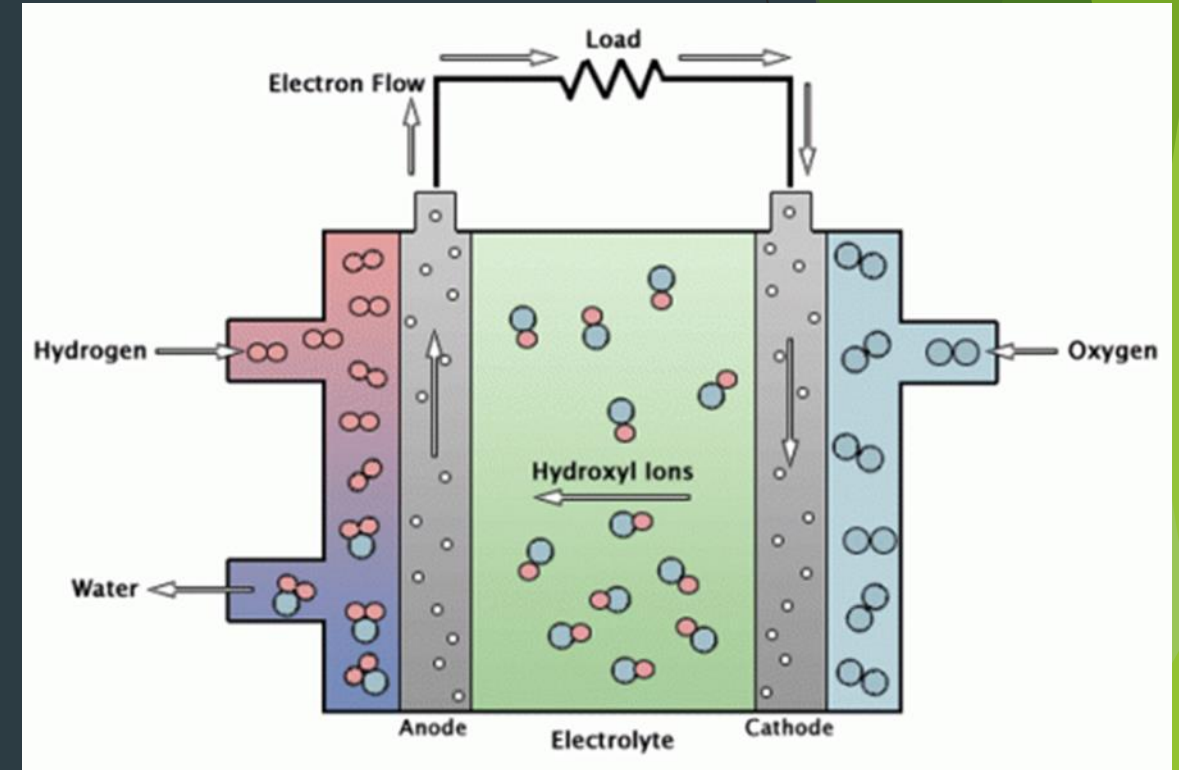


Fig. 1: Fuel Cell Operation

Current Set Up

- ▶ 200 mL Electrolysis Cylinders
- ▶ 1/8 in diameter exit
- ▶ 0.05" Wire to 9V battery
 - ▶ Producing a flow rate of 0.2678L/hr
 - ▶ Fluid Velocity of 0.036m/s
 - ▶ Very low feeding pressure and flow rate
 - ▶ Not optimal to study thermal fluids system



Fig. 2a: Electrolysis Cylinder

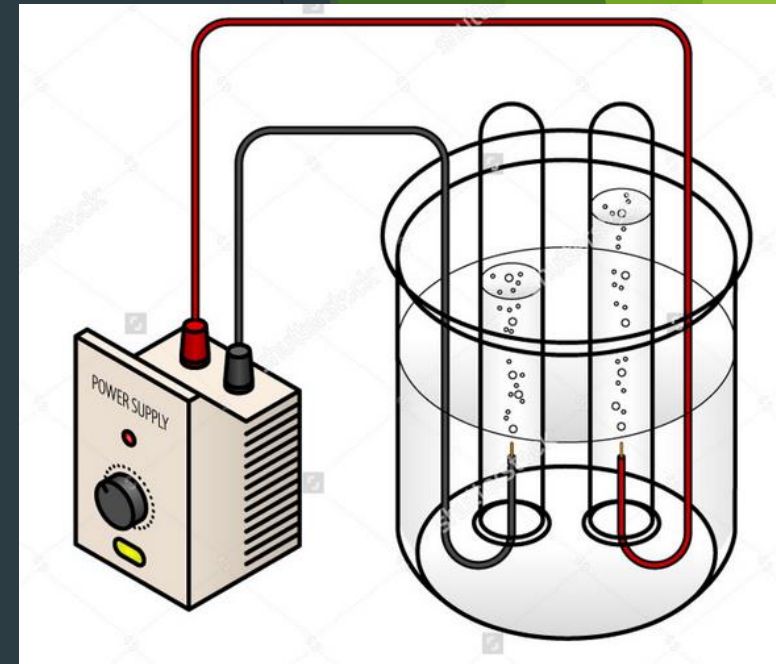


Fig. 2b: Electrolysis Operation

Parallel Configuration

- ▶ Lowest pressure drop
 - ▶ No minor losses
 - ▶ Multi-path design
- ▶ Does not require a high flow rate for operation
- ▶ Current max voltage 0.561V produced
 - ▶ Ideal 1.23V



Fig. 3: Parallel Configuration

Serpentine Configuration

- ▶ Higher pressure drop compared to parallel
 - ▶ Minor losses-22 bends in current design
 - ▶ One path continuous flow
 - ▶ Steady contraction loss
- ▶ Higher flow rate desired for optimal functionality

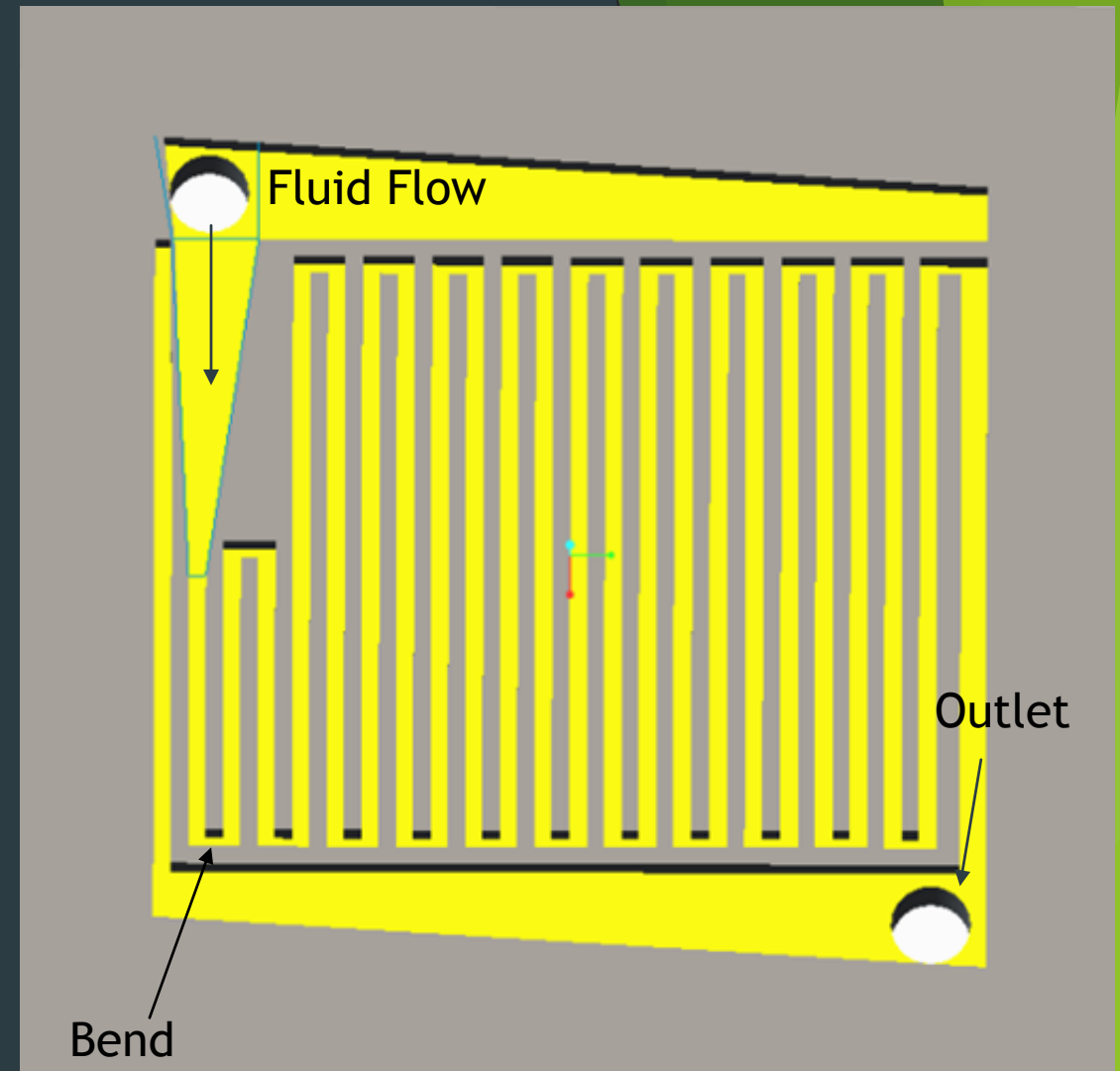


Fig. 4: CAD Serpentine Configuration

Interdigitated Configuration

- ▶ Highest Pressure drop compared to parallel and serpentine design
 - ▶ Minor losses-22 bends
 - ▶ Dead Ends
 - ▶ Forces fluid to lateral diffuse over channel walls for high diffusion rate
 - ▶ Potentially flooding
 - ▶ High pressure drop
- ▶ Higher flow rate required for optimal functionality

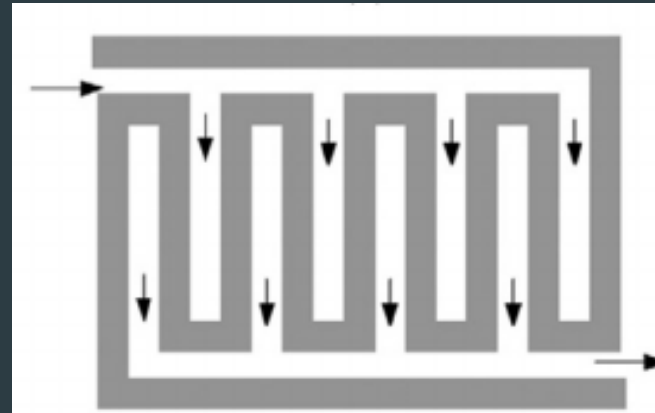


Fig. 5a: Interdigitated Configuration flow

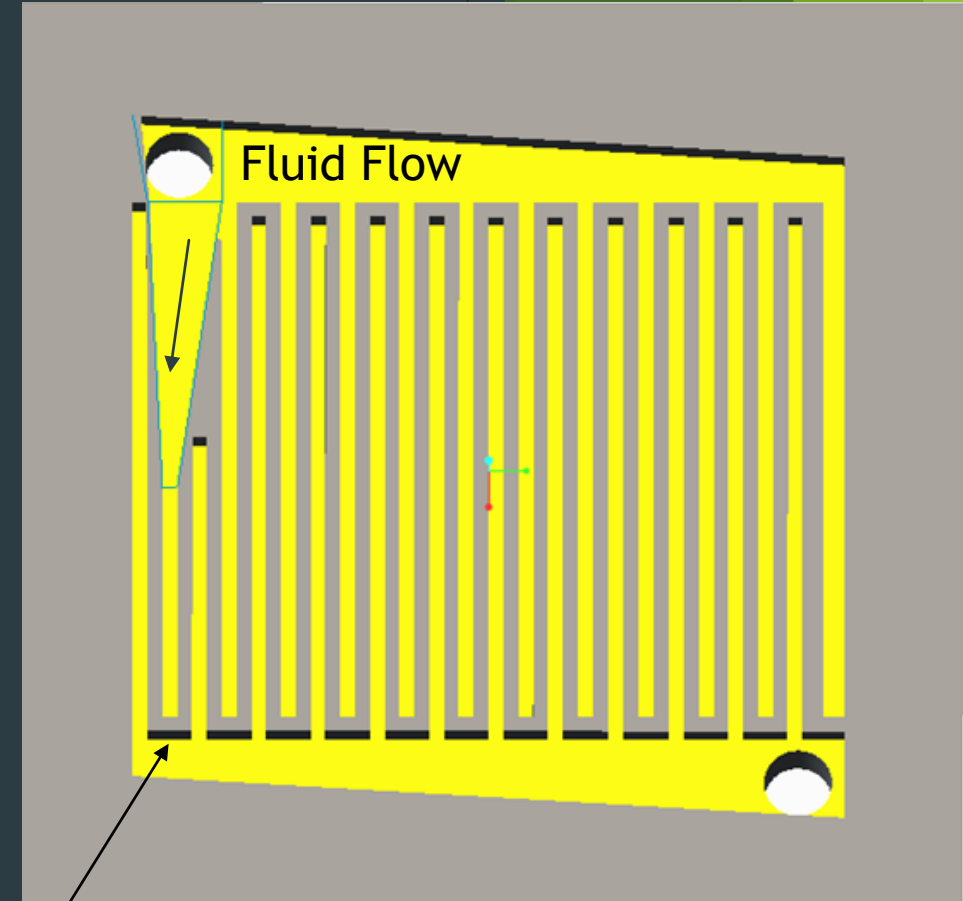


Fig. 5b: CAD Interdigitated Configuration

Dead End

Tristan Walter
Design and Development of AMFC Kit

Effects of Low Flow Rate on High Pressure Drop Configurations

- ▶ Heat distribution and Heat transfer
 - ▶ Fluid temperature will dissipate quicker through system
 - ▶ Max Heat ≈ 393.15 K (Shown in Fig. 6 in red)
- ▶ Why optimize?
 - ▶ Higher flow rates allow for higher heat transfer rates
 - ▶ High current density

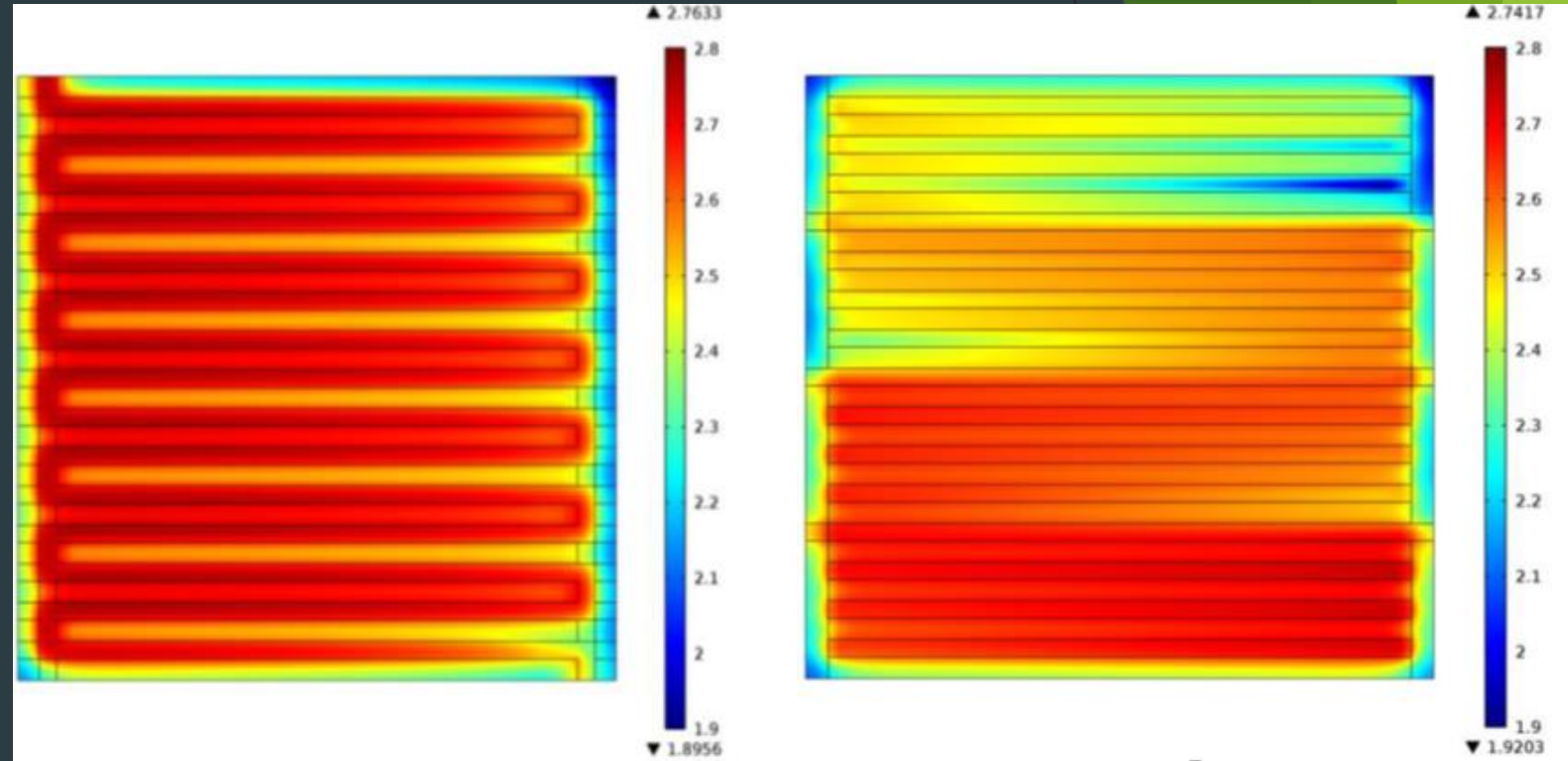


Fig. 6: Thermal Imaging Interdigitated (left) Serpentine (Right)

Effects of Low Flow Rate on High Pressure Drop Configurations

- ▶ Low feeding pressure
 - ▶ Thermal fluids system has energy losses - Head loss
 - ▶ Cannot overcome these losses to allow for fluids to move through system
- ▶ Low Reynolds number
 - ▶ Current set up produces RE=0.42
 - ▶ No empirical formula to find appropriate friction factor
 - ▶ Cannot study pressure drop in the thermal fluid system

$$Re = \frac{\rho v D}{\mu}$$

Eqn. 1: Reynolds Number

$$f = \frac{24}{Re} (1 - 1.3553a + 1.9467a^2 - 1.7012a^3 + 0.9564a^4 - 0.2537a^5)$$

Eqn. 2: R.K. Shah and A.L. London friction factor

Analysis

► How much feeding pressure is required for a respectable flow rate?

► New velocity selected to received respected RE

► 6.0 m/s results in RE = 56.45

► R.K. Shah and A.L. London equation used for f

► Geankoplis table

► Geankoplis friction loss

► $\Delta P = 0.1663 \text{ psi}$

Table 1: Geankoplis table of friction losses for non-tubulent flow

Re	50	100	200	400	1000	Turbulent
K_f	17	7	2.5	1.2	0.85	0.75

$$\sum F = \frac{V^2}{2} \left(\frac{4fL}{D_h} + K_f \right)$$

Eqn. 3: Geankoplis friction loss (Energy Balance)

$$\frac{\Delta P}{\rho} = \sum F$$

Eqn. 4: ΔP is proportional to the sum of frictional losses

Results

- ▶ New flow rate of $1.2384 * 10^{-4} \frac{m^3}{s}$ (Fluid Velocity = 6.0m/s) results in $\Delta P=0.1663\text{psi}$ for current serpentine design
- ▶ Must produce a feeding pressure $> 0.1663\text{psi}$
 - ▶ Increase Electrolysis reaction
 - ▶ Pump

Increasing Pressure and Future Steps

- ▶ Increasing Electrolysis Reaction
 - ▶ Add salt to distilled water in cylinders- increases conductivity
 - ▶ Thicker wire- increases current
 - ▶ Use larger battery- increases voltage
 - ▶ Use pressure gage to measure if this pressure is achievable
- ▶ Pump
 - ▶ Pump selection will allow for an exact desired flow rate
 - ▶ Adds more components to system
 - ▶ Electrolysis method cannot be used



Fig. 7: Pressure Gage used to measure pressure on cylinder

Design for Commercialization

- ▶ Kit's main use is for educational use
 - ▶ Easy to disassemble and replace with different flow plates
 - ▶ Safe design
 - ▶ Maximize simplicity and reliability of the design
 - ▶ Transportability
 - ▶ All parts and directions included

Failure Analysis

Table 2: Failure Analysis for educational kit

#	Name	Failure Mode	Cause	Symptoms and Local Effects	Method of Detection	Effect on System	Remarks and other Effects
1	End Plates	Oxidation, Warped, Damaged	Corrosion, Poor Thermal Management, Neglect	Reduced Diffusion, Leaking, Poor Water Vapor Management	Visual Inspection	Reduced Power Generation	Could pose a health hazard
2	Membrane	Reduced efficiency	Carbon Dioxide Poisoning, Overuse	Uneven Current Distribution	Measuring power output	Reduced Power Generation	Requires Replacement
3	Gas Delivery Tubes	Cracked, Leaking	Dry rot, Loose Connection	Leaking Gas, Reduced Diffusion	Visual Inspection	Reduced Power Generation	Requires Replacement
4	Electrolysis Components	No Gas Production	Dead Battery, Poor Electrical Connections	No Bubbling	Visual Inspection, Testing Battery	No Power Generation	Requires Replacement
5	Electrode Sheets	Salt Build Up, Damaged	Carbon Dioxide Poisoning, Misuse	Uneven Current Distribution, Uneven Heat Distribution	Visual Inspection, Power Output	Reduced Power Generation	Requires Replacement

Safety Analysis

- ▶ AMFC operates at max temperatures of 100°C - 120°C
- ▶ Improving design to make more stable
- ▶ Prevent fuel cell from moving during operation

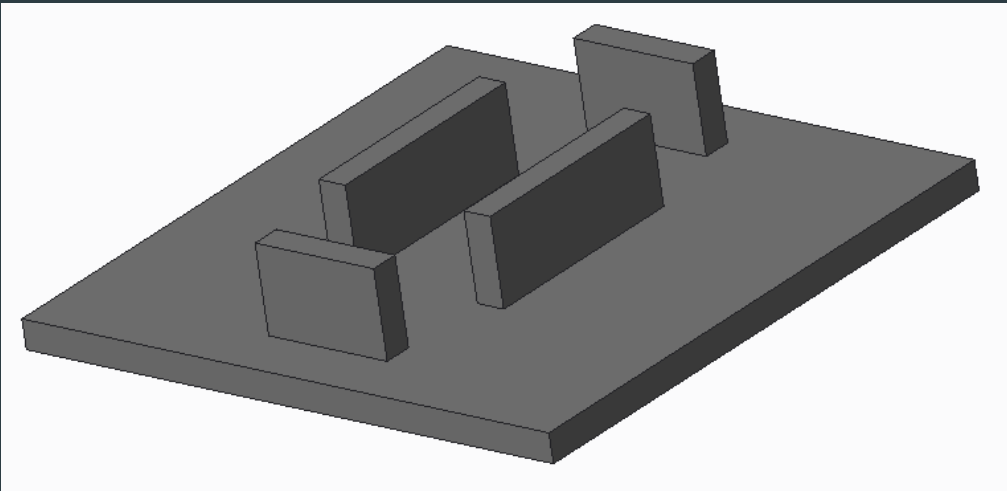


Fig. 8: CAD drawing of fuel cell stand

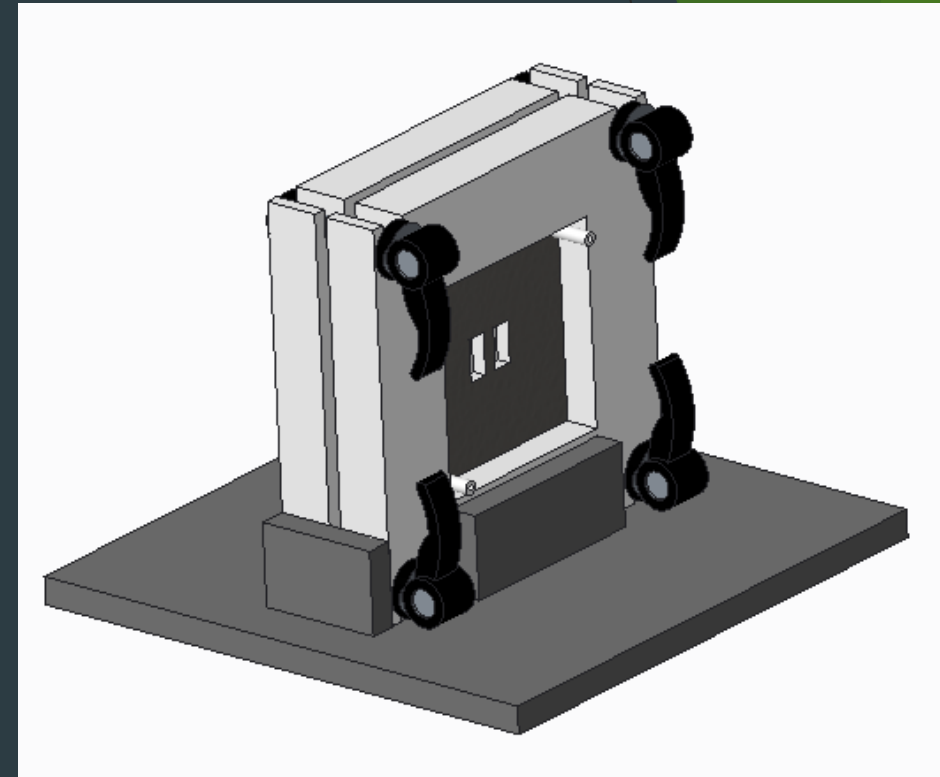


Fig. 9: CAD drawing of fuel cell in stand

Maximizing Simplicity

- ▶ Old design relied on bolts for assembly
- ▶ New design assembled with quick release bicycle skewers
- ▶ Skewers will be customized in order to fit design

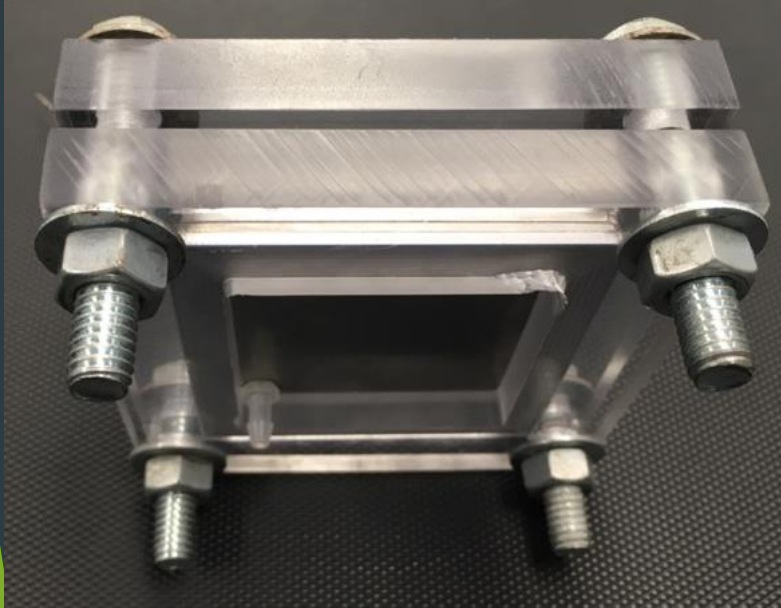


Fig. 10: Current design using bolts



Fig. 11: Quick release skewer

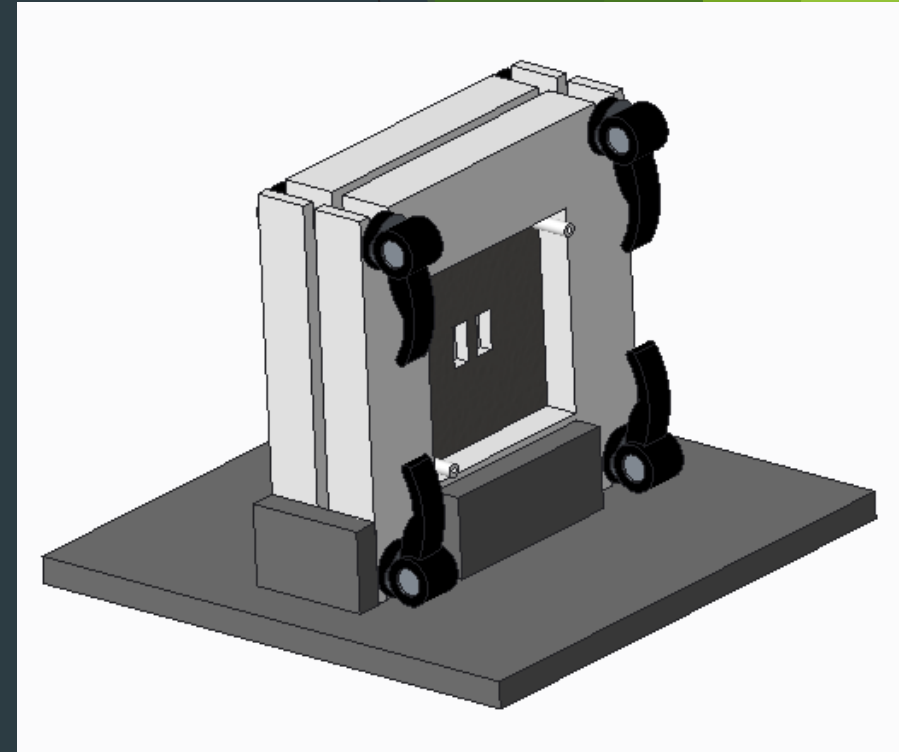


Fig. 12: CAD drawing of new design

Maximizing Testing Reliability

- ▶ Old Design relied on spliced wires to measure power generation
- ▶ Could produce inconsistent measurements
- ▶ New Design features integrated connection points to facilitate alligator clips

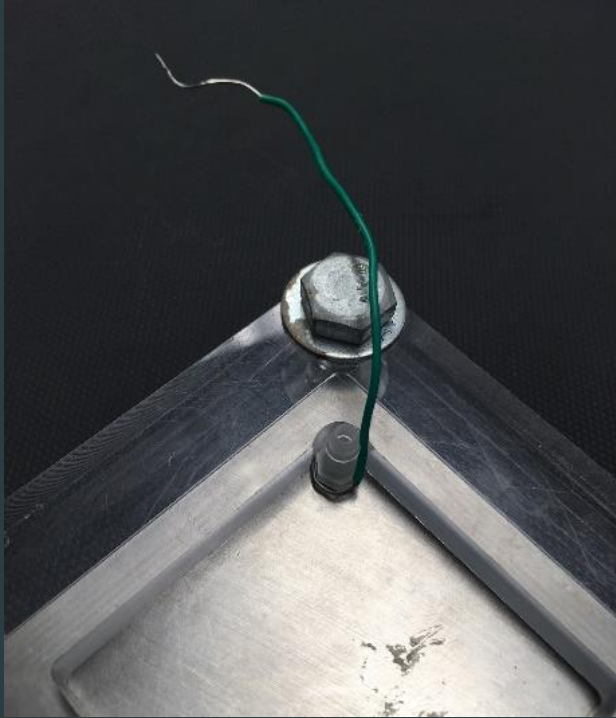


Fig. 13: Current design wiring system

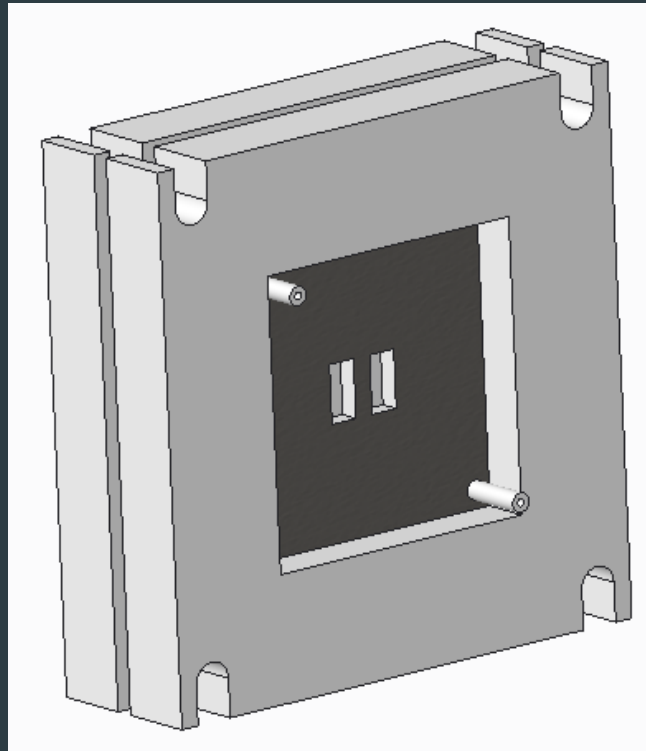


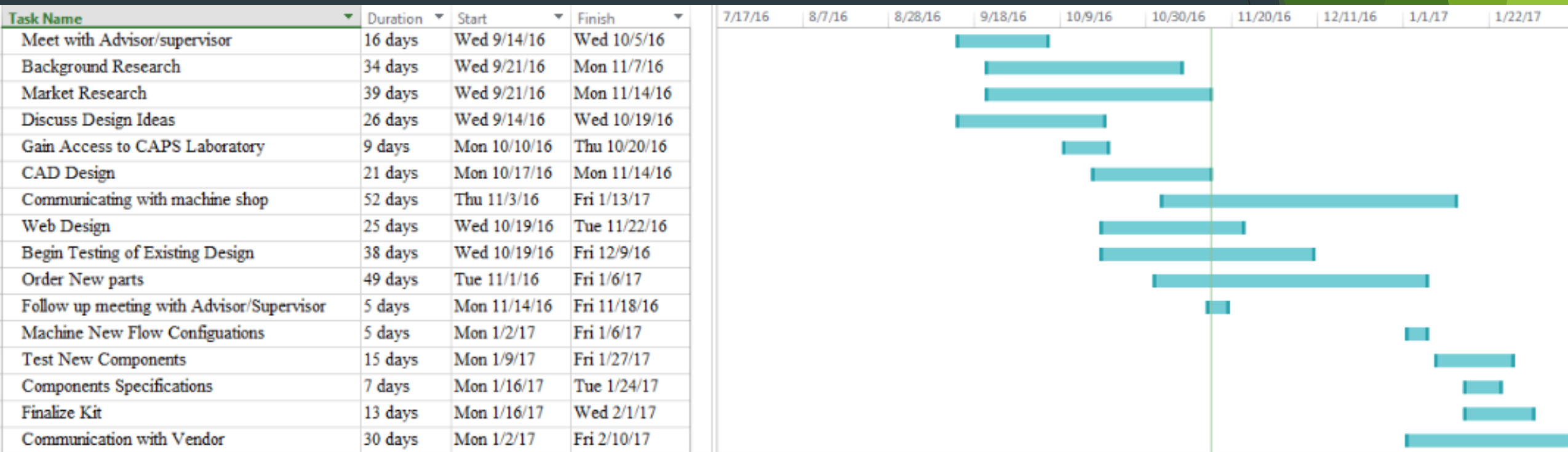
Fig. 14: CAD drawing of end plate modification



Fig. 15: Alligator clips

Gantt Chart

Table 3: Gantt Chart used to organize and plan project



References

1. [4] Sommer, E.M., L.S. Martins, J.V.C. Vargas, J.E.F.C. Gardolinski, J.C. Ordonez, and C.E.B. Marino. "Alkaline Membrane Fuel Cell (AMFC) Modeling and Experimental Validation." *Journal of Power Sowers* (2012): n. pag. Web. 25 Sept. 2016.
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4. Keith, Jason M. *CACHE Modules on Energy in the Curriculum Fuel Cells*. Rep. Houghton: Michigan Technological U, 2009. Print

Questions

